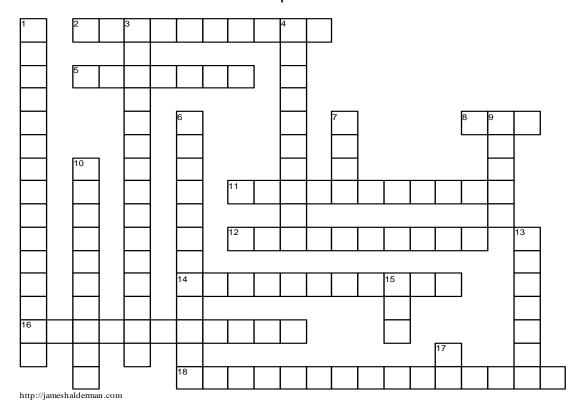
# **Batteries**

# Chapter 50



## ACROSS

# pure \_\_\_\_\_\_. is the release of hydrogen and oxygen from the battery that occurs during charging and results in water usage. Battery sizes are standardized by the \_\_\_\_. Conventional batteries use a liquid electrolyte called \_\_\_\_\_\_ batteries. Each cell is separated from the other cells by \_\_\_\_\_, which are made of the same material as that used for the outside case of the battery. Both types of valve-regulated, lead-acid batteries are also called \_\_\_\_\_\_ battery design.

\_ is the term used to describe the

2 The negative plates are pasted to the grid with a

### **DOWN**

1

1

1

1	Abattery uses little water
	during normal service because of the alloy
	material used to construct the battery plate grids.
3	The rating for batteries is the
	number of minutes that a battery can produce 25
	amperes and still have a battery voltage of 1.75
	volts per cell.
4	is an older battery rating system
	that measures how many amperes of current the
	battery can produce over a period of time.
6	Another name for a maintenance-free battery is a
	battery .
7	Each positive and negative plate in a battery is
	constructed on framework, or, made
	primarily of lead.
9	are constructed of positive and negative
	plates with insulating separators between each
	plate.
0	is a pure porous lead.
3	Another name for a cell is an
5	The acid used in an battery is totally
	absorbed into the separator, making the battery
	leakproof and spillproof.
7	The designation refers to the number of
	amperes that can be supplied by a batter at 32°F
	(0°C).



acid solution in a battery.

volume of water.