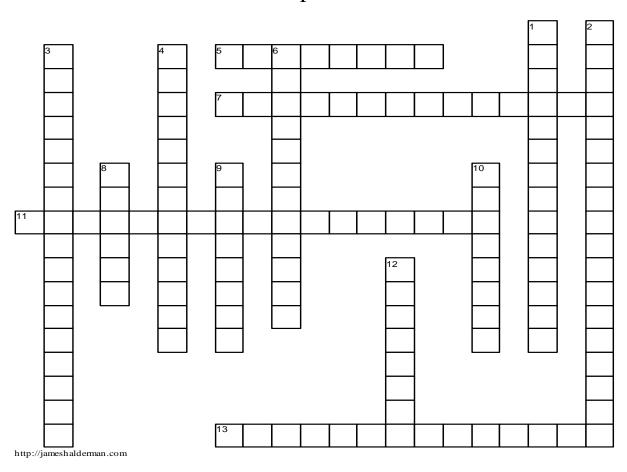
Fuel Cells and Advanced Technologies Chapter 19



ACROSS

5	A is an electrochemical device in which the chemical energy of hydrogen and oxygen is converted
	into electrical energy.
7	Hydrogen is an excellent fuel because it has a very high when compared to an equivalent
	amount of fossil fuel.
11	The part of the PEM fuel cell that contains the membrane, catalyst coatings, and electrodes is known as the
	Assembly (MEA).
13	The fuel-cell design that is best suited for automotive applications is the Membrane (PEM).
DOWN	
1	cells are based on double-layer technology, in which two activated carbon electrodes are
	immersed in an organic electrolyte.
2	The Proton Exchange Membrane fuel cell is also known as a Fuel Cell (PEFC).
3	Compression Ignition (HCCI) is a combustion process.
4	While hydrogen can be used as a fuel, it is not an energy source. Instead, hydrogen is only an
	, as energy must be expended to generate the hydrogen and store it so it can be used as a fuel.
6	The chemical reaction in a fuel cell is the opposite of
8	A fuel-cell vehicle (FCV) uses the fuel cell as its only source of power, whereas a fuel-cell vehicle
	(FCHV) also has an electrical storage device that can be used to power the vehicle.
9	A single fuel cell by itself is not particularly useful, as it generates less than 1 volt of electrical potential. It is more
	common for hundreds of fuel cells to be built together in a stack.
10	A vehicle (FCV) uses the fuel cell as its only source of power, whereas a fuel-cell hybrid vehicle
	(FCHV) also has an electrical storage device that can be used to power the vehicle.
12	One of the major challenges for engineers in this regard is the fact that the heat generated by the fuel cell is
	classified as heat.

