Name_____

SHORT ANSWER. Write the word or phrase that best completes each statement or answers the question.

- 1) What are two designs of NiMH batteries?
- 2) What are the advantages and disadvantages of NiMH batteries?

3) What are two types of high-voltage batteries used in hybrid electric vehicles?

4) What are the advantages and disadvantages of lithium-ion batteries?

5) Why are NiMH batteries are known as alkaline batteries?

- 1) Cylindrical type
 - Prismatic type
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- 2) ADVANTAGES: Nickel-based alkaline batteries have a number of advantages over other battery designs, including the following:
 - High specific energy
 - The nickel electrode can be manufactured with large surface areas, which increase the overall battery capacity
 - The electrolyte does not react with steel, so NiMH batteries can be housed in sealed steel containers that
 - transfer heat reasonably well
 - The materials used in NiMH batteries are environmentally friendly and can be recycled.
 - Excellent cycle life
 - Durable and safe
 - DISADVANTAGES: Disadvantages of the NiMH battery include the following:
 - High rate of self-discharge, especially at elevated temperatures
 - Moderate levels of memory effect, although this seems to be less prominent in newer designs
 - Moderate to high cost
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- 3) Most current production HEVs use nickel-metal hydride (NiMH) battery technology for the high-voltage battery.
 - A battery design that shows a great deal of promise for EVs and HEVs applications is lithiumion (Li-ion) technology
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- 4) ADVANTAGES: Lithium-ion batteries have the following advantages:
 - High specific energy
 - Good high-temperature performance
 - Low self-discharge
 - Minimal memory effect
 - High nominal cell voltage. The nominal voltage of a lithium-ion cell is 3.6 volts, which is three times that of nickel-based alkaline batteries. This allows for fewer battery cells being required to produce high voltage from an HV battery.

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5) NiMH batteries are known as alkaline batteries due to the alkaline (pH greater than 7) nature of the electrolyte Page Ref: 1054